

## Divalproex Sodium Delayed-Release Capsules

<b>Type of Posting</b>	Revision Bulletin
<b>Posting Date</b>	26-Mar-2021
<b>Official Date</b>	1-Apr-2021
<b>Expert Committee</b>	Small Molecules 4

In accordance with the Rules and Procedures of the Council of Experts, the Small Molecules 4 Expert Committee has revised the Divalproex Sodium Delayed-Release Capsules monograph. The purpose for the revision is to add *Dissolution Test 5* to accommodate FDA-approved drug products with different dissolution conditions and/or tolerances than the existing dissolution test(s).

- *Dissolution Test 5* was validated using a Novapak Phenyl brand of L11 column. The typical retention time for valproic acid is about 6 min.

The Divalproex Sodium Delayed-Release Capsules Revision Bulletin supersedes the currently official monograph.

Should you have any questions, please contact Heather Joyce, Senior Scientific Liaison, Team Lead (301-998-6792 or [hri@usp.org](mailto:hri@usp.org)).

## Divalproex Sodium Delayed-Release Capsules

### DEFINITION

Divalproex Sodium Delayed-Release Capsules contain an amount of divalproex sodium equivalent to NLT 90.0% and NMT 110.0% of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ).

### IDENTIFICATION

- **A. [SPECTROSCOPIC IDENTIFICATION TESTS](#) (197), [Infrared Spectroscopy](#): 197K**

**Diluent:** [Acetonitrile](#) and [water](#) (1:1)

**Standard:** Prepare as directed in 197F using [USP Valproic Acid RS](#).

**Sample:** Dissolve the contents of 20 Capsules in 30 mL of *Diluent* in a 50-mL volumetric flask. Sonicate for 30 min to dissolve. Dilute with *Diluent* to volume. Centrifuge the solution at 3000 rpm for about 20 min. Pipet 20 mL of the supernatant into a separatory funnel. Extract with 50 mL of [n-hexane](#). Collect the [n-hexane](#) layer and evaporate the solvent. Cast 1 mg of the liquid obtained after evaporation to sodium chloride (NaCl) windows.

- **B.** The retention time of the major peak of the *Sample solution* corresponds to that of the *Standard solution*, as obtained in the *Assay*.

### ASSAY

#### • PROCEDURE

**Buffer:** 6.8 g/L of [monobasic potassium phosphate](#). Adjust with [phosphoric acid](#) to a pH of 3.0.

**Mobile phase:** [Acetonitrile](#) and *Buffer* (2:3)

**Diluent:** [Acetonitrile](#) and [water](#) (1:1)

**Standard solution:** Transfer a suitable amount of [USP Valproic Acid RS](#) to a suitable volumetric flask to obtain a solution having a final concentration of 2.5 mg/mL of valproic acid. Add 40% of the flask volume of *Diluent*. Sonicate for 5 min and add 20% of the flask volume of [0.1 N hydrochloric acid](#). Dilute with *Diluent* to volume.

**Sample solution:** Transfer an amount of contents (from NLT 20 Capsules) to a suitable volumetric flask to obtain a nominal concentration of 2.5 mg/mL of valproic acid. Dissolve in 20% of the flask volume of [0.1 N hydrochloric acid](#) and sonicate for 5 min. Add 60% of the flask volume of *Diluent* and sonicate for an additional 25 min. Dilute with *Diluent* to volume. Centrifuge at 4000 rpm for 10 min and use the clear supernatant.

#### Chromatographic system

(See [Chromatography](#) (621), [System Suitability](#).)

**Mode:** LC

**Detector:** UV 215 nm

**Column:** 4.6-mm × 15-cm; 5- $\mu$ m packing [L1](#)

**Flow rate:** 1.8 mL/min

**Injection volume:** 20  $\mu$ L

#### System suitability

**Sample:** *Standard solution*

#### Suitability requirements

**Tailing factor:** NMT 2.0 for valproic acid

**Relative standard deviation:** NMT 2.0% for valproic acid

## Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) in the portion of Capsules taken:

$$\text{Result} = (r_U/r_S) \times (C_S/C_U) \times 100$$

$r_U$  = peak response from the *Sample solution*

$r_S$  = peak response from the *Standard solution*

$C_S$  = concentration of [USP Valproic Acid RS](#) in the *Standard solution* (mg/mL)

$C_U$  = nominal concentration of valproic acid in the *Sample solution* (mg/mL)

**Acceptance criteria:** 90.0%–110.0%

## PERFORMANCE TESTS

### Change to read:

- [DISSOLUTION](#) <711>

#### Test 1

**Medium:** Phosphate buffer, pH 7.5 (6.8 g/L of [monobasic potassium phosphate](#) and 1.64 g/L of [sodium hydroxide](#) in [water](#); adjusted with [0.08 N hydrochloric acid TS](#) to a pH of 7.5); 500 mL, degassed

**Apparatus 2:** 50 rpm, with sinkers

**Times:** 2, 4, and 6 h

**Buffer and Mobile phase:** Prepare as directed in the Assay.

**Standard stock solution:** 1.6 mg/mL of [USP Valproic Acid RS](#) in *Mobile phase*

**Standard solution:** 0.26 mg/mL of valproic acid from the *Standard stock solution* and *Medium*

**Sample solution:** Pass a portion of the solution under test through a suitable filter of 0.45- $\mu$ m pore size. Replace the volume withdrawn with an equal volume of *Medium* previously heated at  $37.0 \pm 0.5^\circ$ .

#### Chromatographic system

(See [Chromatography](#) <621>, [System Suitability](#).)

**Mode:** LC

**Detector:** UV 210 nm

**Column:** 4.6-mm  $\times$  15-cm; 5- $\mu$ m packing [L1](#)

**Flow rate:** 1.8 mL/min

**Injection volume:** 40  $\mu$ L

#### System suitability

**Sample:** *Standard solution*

#### Suitability requirements

**Tailing factor:** NMT 2.0 for valproic acid

**Relative standard deviation:** NMT 2.0% for valproic acid

## Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the concentration ( $C_j$ ) of valproic acid ( $C_8H_{16}O_2$ ) in the sample withdrawn from the vessel at each time point ( $i$ ):

$$\text{Result}_i = (r_U/r_S) \times C_S$$

$r_U$  = peak response from the *Sample solution*

$r_S$  = peak response from the *Standard solution*

$C_S$  = concentration of the *Standard solution* (mg/mL)

Calculate the percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) dissolved at each time point ( $i$ ):

$$\text{Result}_1 = C_1 \times V \times (1/L) \times 100$$

$$\text{Result}_2 = [(C_2 \times V) + (C_1 \times V_S)] \times (1/L) \times 100$$

$$\text{Result}_3 = \{(C_3 \times V) + [(C_2 + C_1) \times V_S]\} \times (1/L) \times 100$$

$C_i$  = concentration of valproic acid in the portion of sample withdrawn at the specified time point (mg/mL)

$V$  = volume of *Medium*, 500 mL

$L$  = label claim (mg/Capsule)

$V_S$  = volume of the *Sample solution* withdrawn at each time point and replaced with *Medium* (mL)

**Tolerances:** See [Table 1](#).

**Table 1**

Time Point ( $i$ )	Time (h)	Amount Dissolved (%)
1	2	15–40
2	4	70–90
3	6	NLT 85

The percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) dissolved at each time point conforms to [Dissolution](#) (711), [Acceptance Table 2](#).

**Test 2:** If the product complies with this test, the labeling indicates that the product meets USP *Dissolution Test 2*.

**Procedure A**

**Medium:** [0.05 M phosphate buffer, pH 7.5](#) (6.8 g/L of [monobasic potassium phosphate](#) and 1.64 g/L of [sodium hydroxide](#) in [water](#); adjusted with [2 N sodium hydroxide](#) to a pH of 7.5); 500 mL

**Apparatus 2:** 50 rpm, contents of the Capsule

**Time:** 15 min

**Standard solution A:** 0.036 mg/mL of [USP Valproic Acid RS](#) in *Medium*. A volume of acetonitrile not exceeding 10% of the total volume may be used to dissolve the valproic acid.

**Sample solution A:** Pass a portion of the solution under test through a suitable filter of 0.45- $\mu$ m pore size.

**Procedure B**

**Medium:** [0.05 M phosphate buffer, pH 7.5](#) (6.8 g/L of [monobasic potassium phosphate](#) and 1.64 g/L of [sodium hydroxide](#) in [water](#); adjusted with [2 N sodium hydroxide](#) to a pH of 7.5); 900 mL

**Apparatus 2:** 50 rpm, with wire helix sinkers

**Time:** 4 h

**Buffer A:** 0.5 g/L of [citric acid](#) and 0.4 g/L of [dibasic sodium phosphate](#) in [water](#)

**Buffer B:** 6.8 g/L of [monobasic potassium phosphate](#) and 1.7 g/L of [sodium hydroxide](#) in [water](#); adjusted with [phosphoric acid](#) to a pH of 7.4

**Mobile phase:** Acetonitrile, *Buffer A*, and *Buffer B* (30:35:35); adjusted with [phosphoric acid](#) to a pH of 3.0

**Standard solution B:** 0.13 mg/mL of [USP Valproic Acid RS](#) in *Medium*. A volume of acetonitrile not exceeding 10% of the total volume may be used to dissolve the valproic acid.

**Sample solution B:** Pass a portion of the solution under test through a suitable filter of 0.45- $\mu$ m pore size.

### Chromatographic system

(See [Chromatography <621>](#), [System Suitability](#).)

**Mode:** LC

**Detector:** UV 210 nm

**Column:** 3.9-mm  $\times$  15-cm; 4- $\mu$ m packing [L11](#)

**Flow rate:** 1.2 mL/min

**Injection volume:** 200  $\mu$ L for *Standard solution A* and *Sample solution A*; 50  $\mu$ L for *Standard solution B* and *Sample solution B*

### System suitability

**Sample:** *Standard solution B*

#### Suitability requirements

**Tailing factor:** NMT 2.0 for valproic acid

**Relative standard deviation:** NMT 2.0% for valproic acid

### Analysis

**Samples:** *Standard solution A*, *Sample solution A*, *Standard solution B*, and *Sample solution B*

Calculate the percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) dissolved at each time point:

$$\text{Result} = (r_U/r_S) \times (C_S/L) \times V \times 100$$

$r_U$  = peak response from *Sample solution A* or *Sample solution B*

$r_S$  = peak response from *Standard solution A* or *Standard solution B*

$C_S$  = concentration of *Standard solution A* or *Standard solution B* (mg/mL)

$L$  = label claim (mg/Capsule)

$V$  = volume of *Medium*; 500 mL for *Sample solution A*, 900 mL for *Sample solution B*

**Tolerances:** NMT 20% of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) is dissolved in 15 min (*Sample solution A*); NLT 80% (Q) of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) is dissolved in 4 h (*Sample solution B*). The percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) dissolved at 4 h conforms to [Dissolution <711>](#), [Acceptance Table 1](#).

**Test 3:** If the product complies with this test, the labeling indicates that the product meets USP *Dissolution Test 3*.

### Medium

**Acid stage medium:** [0.08 N hydrochloric acid TS](#); 900 mL

**Buffer stage medium:** Phosphate buffer, pH 7.5 (6.8 g/L of [monobasic potassium phosphate](#) and 1.6 g/L of [sodium hydroxide](#) in [water](#), prepared as follows. Transfer suitable quantities of [monobasic potassium phosphate](#) and [sodium hydroxide](#) to a suitable volumetric flask. Dissolve in 83% of the flask volume of [water](#) and adjust with [0.1 N hydrochloric acid](#), if necessary, to a pH of 7.5. Dilute the resulting solution with [water](#) to volume.); 900 mL

#### Times

**Acid stage:** 2 h

**Buffer stage:** 4 h

**Apparatus 2:** 50 rpm, with sinkers

**Buffer:** 0.25 g/L of [citric acid](#), 0.2 g/L of [anhydrous dibasic sodium phosphate](#), 3.4 g/L of [monobasic potassium phosphate](#), and 0.85 g/L of [sodium hydroxide](#) in [water](#)

**Mobile phase:** [Acetonitrile](#) and *Buffer* (45:55); mixed, degassed, and adjusted with [phosphoric acid](#) to a pH of 2.5

**Standard solution:** 0.14 mg/mL of [USP Valproic Acid RS](#) prepared as follows. Transfer a portion of [USP Valproic Acid RS](#) to a suitable volumetric flask. Dissolve in methanol using 5.0% of the final volume. Dilute with *Buffer stage medium* to final volume and mix.

#### Sample solutions

**Acid stage sample solution:** Pass a portion of the solution under test through a suitable filter of 0.45- $\mu$ m pore size, discarding the first 3 mL of filtrate.

**Buffer stage sample solution:** Pass a portion of the solution under test through a suitable filter of 0.45- $\mu$ m pore size, discarding the first 3 mL of filtrate.

#### Chromatographic system

(See [Chromatography \(621\)](#), [System Suitability](#).)

**Mode:** LC

**Detector:** UV 210 nm

**Column:** 3.9-mm  $\times$  15-cm; 4- $\mu$ m packing [L11](#)

**Flow rate:** 1 mL/min

**Injection volume:** 50  $\mu$ L

#### System suitability

**Sample:** *Standard solution*

#### Suitability requirements

**Tailing factor:** NMT 2.0

**Relative standard deviation:** NMT 2.0%

#### Analysis

**Samples:** *Standard solution*, *Acid stage sample solution*, and *Buffer stage sample solution*

Calculate the percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) dissolved at each time point:

$$\text{Result} = (r_U/r_S) \times (C_S/L) \times V \times 100$$

$r_U$  = peak response from the *Acid stage sample solution* or the *Buffer stage sample solution*

$r_S$  = peak response from the *Standard solution*

$C_S$  = concentration of the *Standard solution* (mg/mL)

$L$  = label claim (mg/Capsule)

$V$  = volume of the *Acid stage medium* or the *Buffer stage medium*, 900 mL

**Tolerances:** The requirements for the *Acid stage* and the *Buffer stage* must be met.

**Acid stage:** NMT 30% (*Q*) of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) is dissolved in 2 h (*Acid stage sample solution*). The percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) dissolved at 2 h conforms to [Table 2](#).

**Table 2**

Level	Number Tested	Criteria
$A_1$	6	No individual value exceeds <i>Q</i> dissolved.
$A_2$	6	Average of the 12 units ( $A_1 + A_2$ ) is NMT <i>Q</i> dissolved; and no individual unit is greater than <i>Q</i> + 15% dissolved.
$A_3$	12	Average of the 24 units ( $A_1 + A_2 + A_3$ ) is NMT <i>Q</i> dissolved; and no individual unit is greater than <i>Q</i> + 15% dissolved.

**Buffer stage:** NLT 80% (*Q*) of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) is dissolved in 4 h (*Buffer stage sample solution*). The percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) dissolved at 4 h conforms to [Dissolution \(711\)](#), [Acceptance Table 2](#).

**Test 4:** If the product complies with this test, the labeling indicates that the product meets USP *Dissolution Test 4*.

**Medium:** [0.05 M phosphate buffer, pH 7.5](#) (6.8 g/L of [monobasic potassium phosphate](#) in [water](#); adjusted with [2 N sodium hydroxide](#) to a pH of 7.5); 500 mL

**Apparatus 2:** 50 rpm

**Times:** 2, 4, and 8 h

**Buffer A:** 0.5 g/L of [citric acid](#) and 4 g/L of [dibasic sodium phosphate](#) in [water](#)

**Buffer B:** 6.8 g/L of [monobasic potassium phosphate](#) and 1.7 g/L of [sodium hydroxide](#) in [water](#); adjusted with [phosphoric acid](#) to a pH of 7.4

**Mobile phase:** [Acetonitrile](#), *Buffer A*, and *Buffer B* (30:35:35); adjusted with [phosphoric acid](#) to a pH of 3.0

**Standard solution:** 0.25 mg/mL of [USP Valproic Acid RS](#) in *Medium*

**Sample solution:** Pass a portion of the solution under test through a suitable filter of 0.45- $\mu$ m pore size.

#### Chromatographic system

(See [Chromatography \(621\)](#), [System Suitability](#).)

**Mode:** LC

**Detector:** UV 210 nm

**Column:** 3.9-mm  $\times$  15-cm; 4- $\mu$ m packing [L11](#)

**Column temperature:** 30°

**Flow rate:** 1.2 mL/min

**Injection volume:** 50  $\mu$ L

**Run time:** NLT 1.5 times the retention time of valproic acid

#### System suitability

**Sample:** *Standard solution*

**Suitability requirements**

**Tailing factor:** NMT 2.0 for valproic acid

**Relative standard deviation:** NMT 2.0% for valproic acid

**Analysis**

**Samples:** *Standard solution* and *Sample solution*

Calculate the concentration ( $C_i$ ) of valproic acid ( $C_8H_{16}O_2$ ) in the sample withdrawn from the vessel at each time point ( $i$ ):

$$\text{Result}_i = (r_U/r_S) \times C_S$$

$r_U$  = peak response from the *Sample solution*

$r_S$  = peak response from the *Standard solution*

$C_S$  = concentration of the *Standard solution* (mg/mL)

Calculate the percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) dissolved at each time point ( $i$ ):

$$\text{Result}_1 = C_1 \times V \times (1/L) \times 100$$

$$\text{Result}_2 = \{[C_2 \times (V - V_S)] + (C_1 \times V_S)\} \times (1/L) \times 100$$

$$\text{Result}_3 = \{[C_3 \times [V - (2 \times V_S)]] + [(C_2 + C_1) \times V_S]\} \times (1/L) \times 100$$

$C_i$  = concentration of valproic acid in the portion of sample withdrawn at the specified time point (mg/mL)

$V$  = volume of *Medium*, 500 mL

$L$  = label claim (mg/Capsule)

$V_S$  = volume of the *Sample solution* withdrawn at each time point (mL)

**Tolerances:** See [Table 3](#).

**Table 3**

Time Point ( $i$ )	Time (h)	Amount Dissolved (NLT %)
1	2	60
2	4	70
3	8	80

The percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) dissolved at each time point conforms to [Dissolution \(711\)](#), [Acceptance Table 4](#).

**▲ Test 5:** If the product complies with this test, the labeling indicates that the product meets USP [Dissolution Test 5](#).

**Medium:** 0.05 M phosphate buffer, pH 7.5 (6.8 g/L of [monobasic potassium phosphate](#) and 1.64 g/L of [sodium hydroxide](#) in water; adjusted with [2 N sodium hydroxide](#) to a pH of 7.5); 900 mL, deaerated

**Apparatus 2:** 50 rpm with suitable sinkers



**Times:** 1 and 4 h

**Buffer A:** 0.5 g/L of citric acid and 0.4 g/L of dibasic sodium phosphate in water

**Buffer B:** 6.8 g/L of monobasic potassium phosphate and 1.7 g/L of sodium hydroxide in water; adjusted with phosphoric acid to a pH of 7.4

**Mobile phase:** Acetonitrile, Buffer A, and Buffer B (30:35:35); adjusted with phosphoric acid to a pH of 3.0

**Standard solution:**  $(L/900)$  mg/mL of USP Valproic Acid RS in Medium, where  $L$  is the label claim in mg/Capsule

**Sample solution:** Pass a portion of the solution under test through a suitable filter of 0.45- $\mu$ m pore size, discarding the first 2–3 mL of the filtrate.

### Chromatographic system

(See Chromatography <621>, System Suitability.)

**Mode:** LC

**Detector:** UV 210 nm

**Column:** 3.9-mm  $\times$  15-cm; 4- $\mu$ m packing L11

**Flow rate:** 1.2 mL/min

**Injection volume:** 50  $\mu$ L

**Run time:** NLT 1.5 times the retention time of valproic acid

### System suitability

**Sample:** *Standard solution*

#### Suitability requirements

**Tailing factor:** NMT 2.0 for valproic acid

**Relative standard deviation:** NMT 2.0% for valproic acid

### Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the concentration ( $C_i$ ) of valproic acid ( $C_8H_{16}O_2$ ) in the sample withdrawn from the vessel at each time point ( $i$ ):

$$\text{Result}_i = (r_U/r_S) \times C_S$$

$r_U$  = peak response from the *Sample solution*

$r_S$  = peak response from the *Standard solution*

$C_S$  = concentration of the *Standard solution* (mg/mL)

Calculate the percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) dissolved at each time point ( $i$ ):

$$\text{Result}_1 = C_1 \times V \times (1/L) \times 100$$

$$\text{Result}_2 = \{[C_2 \times (V - V_S)] + (C_1 \times V_S)\} \times (1/L) \times 100$$

$C_i$  = concentration of valproic acid in the portion of sample withdrawn at the specified time point (mg/mL)

$V$  = volume of *Medium*, 900 mL

$L$  = label claim (mg/Capsule)

$V_S$  = volume of the *Sample solution* withdrawn at each time point (mL)

**Tolerances:** See Table 4.

**Table 4**

Time Point ( <i>i</i> )	Time (h)	Amount Dissolved (%)
1	1	NMT 25
2	4	NLT 80

The percentage of the labeled amount of valproic acid ( $C_8H_{16}O_2$ ) dissolved at each time point conforms to *Dissolution* <711>, *Acceptance Table 2*. ▲ (RB 1-Apr-2021)

- **UNIFORMITY OF DOSAGE UNITS** <905>: Meet the requirements

#### **ADDITIONAL REQUIREMENTS**

- **PACKAGING AND STORAGE:** Preserve in tight, light-resistant containers at controlled room temperature.
- **LABELING:** Divalproex Sodium Delayed-Release Capsules may be swallowed whole or may be administered by carefully opening the Capsule and sprinkling the entire contents on a small amount of soft food. This drug/food mixture should be swallowed immediately and not chewed. It should not be stored for future use. When more than one *Dissolution* test is given, the labeling states the *Dissolution* test used only if *Test 1* is not used.
- **USP REFERENCE STANDARDS** <11>.  
[USP Valproic Acid RS](#)

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